ABE Powder Extinguisher with compressed gas

Altamonte
Chemwatch: 5570-64
Version No: 2.1

Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements

Chemwatch Hazard Alert Code: 2

Issue Date: **13/01/2023** Print Date: **13/01/2023** S.GHS.AUS.EN.E

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier		
Product name	ABE Powder Extinguisher with compressed gas	
Chemical Name	Not Applicable	
Synonyms	(FP1021,FP1531,FP2545,FP4566,FP90108,FPA550)	
Proper shipping name	FIRE EXTINGUISHERS with compressed or liquefied gas	
Chemical formula	Not Applicable	
Other means of identification	Not Available	

Relevant identified uses of the substance or mixture and uses advised against

Details of the manufacturer or supplier of the safety data sheet

Registered company name	Altamonte
Address	138-140 Bayfield Rd East Bayswater North VIC 3153 Australia
Telephone	+61 3 9720 4333
Fax	Not Available
Website	http://altamonte.com.au/
Email	info@altamonte.com.au

Emergency telephone number

Association / Organisation	Not Available
Emergency telephone numbers	Not Available
Other emergency telephone numbers	Not Available

SECTION 2 Hazards identification

Classification of the substance or mixture

Poisons Schedule	Not Applicable	
Classification ^[1]	Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 2A, Specific Target Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, Hazardous to the Aquatic Environment Long-Term Hazard Category 4	
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI	

Label elements

Hazard pictogram(s)



Signal word	Warning

Hazard statement(s)

AUH044	Risk of explosion if heated under confinement.
H315	Causes skin irritation.
H319	Causes serious eye irritation.
H335	May cause respiratory irritation.
H413	May cause long lasting harmful effects to aquatic life.

Precautionary statement(s) Prevention

P271	Use only outdoors or in a well-ventilated area.
P261	Avoid breathing dust/fumes.

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P273	Avoid release to the environment.
P280	Wear protective gloves, protective clothing, eye protection and face protection.
P264	Wash all exposed external body areas thoroughly after handling.

Precautionary statement(s) Response

P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P312	P312 Call a POISON CENTER/doctor/physician/first aider/if you feel unwell.	
P337+P313	If eye irritation persists: Get medical advice/attention.	
P302+P352	IF ON SKIN: Wash with plenty of water and soap.	
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.	
P332+P313	If skin irritation occurs: Get medical advice/attention.	
P362+P364	Take off contaminated clothing and wash it before reuse.	

Precautionary statement(s) Storage

P405	Store locked up.
P403+P233	Store in a well-ventilated place. Keep container tightly closed.

Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7722-76-1	80-90	ammonium phosphate, monobasic
7783-20-2	NotSpec	ammonium sulfate
12182-56-8	Balance	staurolite
Not Available		propellant as
7727-37-9.	NotSpec	nitrogen
Legend:	Classified by Chemwatch; 2. Classific Classification drawn from C&L * EU IOL	ation drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. ELVs available

SECTION 4 First aid measures

Description of first aid measures

- Generally not applicable.
- If product comes in contact with eyes remove the patient from gas source or contaminated area.
- ▶ Take the patient to the nearest eye wash, shower or other source of clean water.
- Open the eyelid(s) wide to allow the material to evaporate.
- Gently rinse the affected eye(s) with clean, cool water for at least 15 minutes. Have the patient lie or sit down and tilt the head back. Hold the eyelid(s) open and pour water slowly over the eyeball(s) at the inner corners, letting the water run out of the outer corners.
- The patient may be in great pain and wish to keep the eyes closed. It is important that the material is rinsed from the eyes to prevent further damage.
- Ensure that the patient looks up, and side to side as the eye is rinsed in order to better reach all parts of the eye(s)
- Transport to hospital or doctor.
- Even when no pain persists and vision is good, a doctor should examine the eye as delayed damage may occur.
- If the patient cannot tolerate light, protect the eyes with a clean, loosely tied bandage.
- ► Ensure verbal communication and physical contact with the patient.

DO NOT allow the patient to rub the eyes

DO NOT allow the patient to tightly shut the eyes

DO NOT introduce oil or ointment into the eye(s) without medical advice

DO NOT use hot or tepid water.

Skin Contact

Eye Contact

If skin contact occurs:

- Immediately remove all contaminated clothing, including footwear.
- Flush skin and hair with running water (and soap if available).
- ► Seek medical attention in event of irritation.

▶ Following exposure to gas, remove the patient from the gas source or contaminated area.

- ▶ NOTE: Personal Protective Equipment (PPE), including positive pressure self-contained breathing apparatus may be required to assure the safety of the rescuer.
- Prostheses such as false teeth, which may block the airway, should be removed, where possible, prior to initiating first aid procedures.
- If the patient is not breathing spontaneously, administer rescue breathing.

Inhalation

- If the patient does not have a pulse, administer CPR.
 If medical oxygen and appropriately trained personnel are available, administer 100% oxygen.
- Summon an emergency ambulance. If an ambulance is not available, contact a physician, hospital, or Poison Control Centre for further instruction.
- ▶ Keep the patient warm, comfortable and at rest while awaiting medical care.
- MONITOR THE BREATHING AND PULSE, CONTINUOUSLY.
- Administer rescue breathing (preferably with a demand-valve resuscitator, bag-valve mask-device, or pocket mask as trained) or CPR if necessary.

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Ingestion

Not considered a normal route of entry.

Indication of any immediate medical attention and special treatment needed

For gas exposures:

BASIC TREATMENT

- Festablish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- Monitor and treat, where necessary, for pulmonary oedema .
- Monitor and treat, where necessary, for shock.
- Anticipate seizures.

ADVANCED TREATMENT

- F Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- F Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
- Treat seizures with diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

SECTION 5 Firefighting measures

Extinguishing media

SMALL FIRE: Use extinguishing agent suitable for type of surrounding fire

LARGE FIRE: Cool cylinder.

DO NOT direct water at source of leak or venting safety devices as icing may occur.

Special hazards arising from the substrate or mixture

Fire Incompatibility None known.

Advice for firefighters

Fire Fighting	Slight hazard when exposed to heat, flame and oxidisers.
Fire/Explosion Hazard	 Containers may explode when heated - Ruptured cylinders may rocket Fire exposed containers may vent contents through pressure relief devices. High concentrations of gas may cause asphyxiation without warning. May decompose explosively when heated or involved in fire. Contact with gas may cause burns, severe injury and/ or frostbite. Decomposition may produce toxic fumes of: nitrogen oxides (NOx) phosphorus oxides (NOx) phosphorus oxides (POx) metal oxides Articles and manufactured articles may constitute a fire hazard where polymers form their outer layers or where combustible packaging remains in place. Certain substances, found throughout their construction, may degrade or become volatile when heated to high temperatures. This may create a secondary hazard.
HAZCHEM	Not Applicable

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

- Avoid breathing vapour and any contact with liquid or gas. Protective equipment including respirator should be used.
 DO NOT enter confined spaces where gas may have accumulated.
 Increase ventilation.
 - Minor Spills
 - Clear area of personnel
 - Stop leak only if safe to so do
 - Remove leaking cylinders to safe place. Release pressure under safe controlled conditions by opening valve.
 - Do not exert excessive pressure on the valve; do not attempt to operate a damaged valve
 - Do not exert excessive pressure on the valve; do not attempt to operate a damaged valve;
 Orientate cylinder so that the leak is gas, not liquid, to minimise rate of leakage
 - Keep area clear of personnel until gas has dispersed.

Major Spills

- Clear area of all unprotected personnel and move upwind.
- Alert Emergency Authority and advise them of the location and nature of hazard.
- Wear breathing apparatus and protective gloves.
 Prevent by any means available, spillage from entering drains and water-courses.
- Consider evacuation.Increase ventilation.
- No smoking or naked lights within area.

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- Stop leak only if safe to so do.
- Water spray or fog may be used to disperse vapour.
- ▶ DO NOT enter confined space where gas may have collected.
- ▶ Keep area clear until gas has dispersed.
- Clean up all spills immediately.
- Wear protective clothing, safety glasses, dust mask, gloves.
- Secure load if safe to do so. Bundle/collect recoverable product.
- Use dry clean up procedures and avoid generating dust.
- Vacuum up (consider explosion-proof machines designed to be grounded during storage and use).
- Water may be used to prevent dusting
- Collect remaining material in containers with covers for disposal
- Flush spill area with water.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Safe handling

Precautions for safe handling

- · Consider use in closed pressurised systems, fitted with temperature, pressure and safety relief valves which are vented for safe dispersal. Use only properly specified equipment which is suitable for this product, its supply pressure and temperature
- · The tubing network design connecting gas cylinders to the delivery system should include appropriate pressure indicators and vacuum or
- · Fully-welded types of pressure gauges, where the bourdon tube sensing element is welded to the gauge body, are recommended.
- · Before connecting gas cylinders, ensure manifold is mechanically secure and does not containing another gas. Before disconnecting gas cylinder, isolate supply line segment proximal to cylinder, remove trapped gas in supply line with aid of vacuum pump
- When connecting or replacing cylinders take care to avoid airborne particulates violently ejected when system pressurises.
- · Consider the use of doubly-contained piping; diaphragm or bellows sealed, soft seat valves; backflow prevention devices; flash arrestors; and flow monitoring or limiting devices. Gas cabinets, with appropriate exhaust treatment, are recommended, as is automatic monitoring of the secondary enclosures and work areas for release.
- · Use a pressure reducing regulator when connecting cylinder to lower pressure (<100 psig) piping or systems
- · Use a check valve or trap in the discharge line to prevent hazardous back-flow into the cylinder
- · Check regularly for spills or leaks. Keep valves tightly closed but do not apply extra leverage to hand wheels or cylinder keys.
- · Open valve slowly. If valve is resistant to opening then contact your supervisor
- Valve protection caps must remain in place must remain in place unless container is secured with valve outlet piped to use point.
 - · Never insert a pointed object (e.g hooks) into cylinder cap openings as a means to open cap or move cylinder. Such action can inadvertently turn the valve and gas a gas leak. Use an adjustable strap instead of wrench to free an over-tight or rusted cap.
 - · A bubble of gas may buildup behind the outlet dust cap during transportation, after prolonged storage, due to defective cylinder valve or if a dust cap is inserted without adequate evacuation of gas from the line. When loosening dust cap, preferably stand cylinder in a suitable enclosure and take cap off slowly. Never face the dust cap directly when removing it; point cap away from any personnel or any object that may pose a hazard. under negative pressure (relative to atmospheric gas)
 - Suck back of water into the container must be prevented. Do not allow backfeed into the container.
 - · Do NOT drag, slide or roll cylinders use a suitable hand truck for cylinder movement
 - · Test for leakage with brush and detergent NEVER use a naked flame
 - linder by any means to increase the discharge rate of product from cylinder.
 - · Leaking gland nuts may be tightened if necessary.
 - · If a cylinder valve will not close completely, remove the cylinder to a well ventilated location (e.g. outside) and, when empty, tag as FAULTY and return to supplier.
 - · Obtain a work permit before attempting any repairs.
 - DO NOT attempt repair work on lines,
 - · Atmospheres must be tested and O.K. before work resumes after leakage.
 - DO NOT transfer gas from one cylinder to another.
 - Cylinders should be stored in a purpose-built compound with good ventilation, preferably in the open.
 - Such compounds should be sited and built in accordance with statutory requirements The storage compound should be kept clear and access restricted to authorised personnel only.
 - Cylinders stored in the open should be protected against rust and extremes of weather.
 - Cylinders in storage should be properly secured to prevent toppling or rolling.
 - Cylinder valves should be closed when not in use.
 - Where cylinders are fitted with valve protection this should be in place and properly secured.
 - Gas cylinders should be segregated according to the requirements of the Dangerous Goods Act.
 - Preferably store full and empty cylinders separately.
 - Check storage areas for hazardous concentrations of gases prior to entry.
 - Full cylinders should be arranged so that the oldest stock is used first.
 - Cylinders in storage should be checked periodically for general condition and leakage.
 - Protect cylinders against physical damage. Move and store cylinders correctly as instructed for their manual handling.

NOTE: A 'G' size cylinder is usually too heavy for an inexperienced operator to raise or lower.

Store away from incompatible materials.

Conditions for safe storage, including any incompatibilities

Cvlinder:

- Ensure the use of equipment rated for cylinder pressure.
- Ensure the use of compatible materials of construction.

Suitable container

Other information

- Valve protection cap to be in place until cylinder is secured, connected.
- Cylinder must be properly secured either in use or in storage
- Cylinder valve must be closed when not in use or when empty.
- Segregate full from empty cylinders.

WARNING: Suckback into cylinder may result in rupture. Use back-flow preventive device in piping.

Storage incompatibility

- Avoid strong bases.
- Compressed gases may contain a large amount of kinetic energy over and above that potentially available from the energy of reaction produced by the gas in chemical reaction with other substances

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SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Not Available

Emergency Limits

Ingredient	TEEL-1	TEEL-2	TEEL-3
ammonium phosphate, monobasic	17 mg/m3	190 mg/m3	1,100 mg/m3
ammonium sulfate	13 mg/m3	140 mg/m3	840 mg/m3
nitrogen	7.96E+05 ppm	8.32E+05 ppm	8.69E+05 ppm

Ingredient	Original IDLH	Revised IDLH
ammonium phosphate, monobasic	Not Available	Not Available
ammonium sulfate	Not Available	Not Available
staurolite	Not Available	Not Available
nitrogen	Not Available	Not Available

Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit
ammonium phosphate, monobasic	Е	≤ 0.01 mg/m³
ammonium sulfate	E	≤ 0.01 mg/m³
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a	

Exposure controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

range of exposure concentrations that are expected to protect worker health.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

- Areas where cylinders are stored require good ventilation and, if enclosed, need discrete/controlled exhaust ventilation.
- Secondary containment and exhaust gas treatment may be required by certain jurisdictions.
- Local exhaust ventilation may be required in work areas.
- Consideration should be given to the use of diaphragm or bellows-sealed, soft-seat valves; backflow prevention devices and flow-monitoring or limiting devices.
- Automated alerting systems with automatic shutdown of gas-flow may be appropriate and may in fact be mandatory in certain jurisdictions.
- Respiratory protection in the form of air-supplied or self-contained breathing equipment must be worn if the oxygen concentration in the workplace air is less than 19%.
- Cartridge respirators do NOT give protection and may result in rapid suffocation.

Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Type of Contaminant:	Air Speed:
gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2.5 m/s (200-500 f/min.) for extraction of gases discharged 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

Personal protection

Appropriate engineering

controls











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Safety glasses with side shields. Chemical goggles. ▶ Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent] Eye and face protection Close fitting gas tight goggles **DO NOT** wear contact lenses Londact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lens or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean $environment \ only \ after \ workers \ have \ washed \ hands \ thoroughly. \ [CDC\ NIOSH\ Current \ Intelligence\ Bulletin\ 59], \ [AS/NZS\ 1336\ or\ national\ NZS\ 1336$ equivalent] Skin protection See Hand protection below Wear chemical protective gloves, e.g. PVC. Hands/feet protection Wear safety footwear. **Body protection** See Other protection below Other protection 7252

Respiratory protection

Full face respirator with supplied air.

- Positive pressure, full face, air-supplied breathing apparatus should be used for work in enclosed spaces if a leak is suspected or the primary containment is to be opened (e.g. for a cylinder change)
- Air-supplied breathing apparatus is required where release of gas from primary containment is either suspected or demonstrated.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Blue powder with no odour; Insoluble in water.		
Physical state	Manufactured	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable
pH (as supplied)	4.5	Decomposition temperature (°C)	190
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Applicable
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Applicable	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Mutagenicity

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The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by sleepiness, reduced alertness, loss of reflexes, lack of co-ordination, and vertigo. Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual. Inhaled Inhalation of non-toxic gases may cause: CNS effects: headache, confusion, dizziness, stupor, seizures and coma; respiratory: shortness of breath and rapid breathing: cardiovascular: collapse and irregular heart beats; gastrointestinal: mucous membrane irritation, nausea and vomiting Accidental ingestion of the material may be damaging to the health of the individual. Ingestion Not normally a hazard due to physical form of product Considered an unlikely route of entry in commercial/industrial environments This material can cause inflammation of the skin on contact in some persons. The material may accentuate any pre-existing dermatitis condition **Skin Contact** Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. Eye This material can cause eye irritation and damage in some persons. Not considered to be a risk because of the extreme volatility of the gas. Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems. Chronic Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. Main route of exposure to the gas in the workplace is by inhalation. TOXICITY IRRITATION ABE Powder Extinguisher with compressed gas Not Available Not Available TOXICITY IRRITATION dermal (rat) LD50: >5000 mg/kg[1] Eye: no adverse effect observed (not irritating)[1] ammonium phosphate. monobasic Inhalation(Rat) LC50: >5 mg/l4h[1] Skin: no adverse effect observed (not irritating)^[1] Oral (Rat) LD50; >2000 mg/kg[1] TOXICITY IRRITATION dermal (rat) LD50: >2000 mg/kg[1] Eye: no adverse effect observed (not irritating)[1]ammonium sulfate Oral (Rat) LD50; 2840 mg/kg^[2] Skin: no adverse effect observed (not irritating)[1] TOXICITY IRRITATION staurolite Not Available Not Available TOXICITY IRRITATION nitrogen Not Available Not Available Legend: 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2. Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances For ammonium sulfate: Acute toxicity: Ammonium sulfate has relatively low acute toxicity. In healthy humans, inhaling high concentrations caused mild reduction in lung function. Animal testing has not shown ammonium sulfate to cause irritation to the skin and eyes. There is no available data on sensitization. Repeat dose toxicity: Testing in animals has not shown any chronic toxic effects, except for diarrhoea **AMMONIUM SULFATE** Reproductive toxicity: There are no valid studies available on the effects of ammonium sulfate on fertility and development. Developmental toxicity: Studies of developmental toxicity for ammonium sulfate are not available Genetic toxicity: Testing with bacteria and yeasts did not reveal ammonium sulfate to cause mutations, and in mammalian and human cell cultures, it did not cause chromosomal aberrations. Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria for diagnosing RADS include the absence of previous airways disease in a non-atopic individual, with sudden onset of persistent AMMONIUM PHOSPHATE. asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Other criteria for diagnosis of RADS include a reversible **MONOBASIC & AMMONIUM** airflow pattern on lung function tests, moderate to severe bronchial hyperreactivity on methacholine challenge testing, and the lack of minimal **SULFATE** lymphocytic inflammation, without eosinophilia. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. On the other hand, industrial bronchitis is a disorder that occurs as a result of exposure due to high concentrations of irritating substance (often particles) and is completely reversible after exposure ceases. The disorder is characterized by difficulty breathing, cough and mucus production. AMMONIUM PHOSPHATE. **MONOBASIC & STAUROLITE** No significant acute toxicological data identified in literature search. & NITROGEN **Acute Toxicity** Carcinogenicity Skin Irritation/Corrosion × Reproductivity V Serious Eye Damage/Irritation STOT - Single Exposure V Respiratory or Skin × STOT - Repeated Exposure × sensitisation

Legend:

🗶 – Data either not available or does not fill the criteria for classification

Data available to make classification

Aspiration Hazard

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SECTION 12 Ecological information

Toxicity

DE Bassalan Estinassiahan	Endpoint	Test Duration (hr)	Species	Value	Source
ABE Powder Extinguisher with compressed gas	Not Available	Not Available	Not Available	Not Available	Not Availabl
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	NOEC(ECx)	72h	Algae or other aquatic plants	100mg/l	2
ammonium phosphate, monobasic	EC50	72h	Algae or other aquatic plants	>100mg/l	2
monobasic	EC50	48h	Crustacea	>100mg/l	2
	LC50	96h	Fish	>100mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50(ECx)	24h	Fish	0.068mg/L	5
ammonium sulfate	EC50	72h	Algae or other aquatic plants	190mg/l	2
	EC50	48h	Crustacea	60mg/l	2
	LC50	96h	Fish	34.6mg/l	2
staurolite	Endpoint	Test Duration (hr)	Species	Value	Source
	Not Available	Not Available	Not Available	Not Available	Not Availabl
nitrogen	Endpoint	Test Duration (hr)	Species	Value	Source
	Not Available	Not Available	Not Available	Not Available	Not Availabl
nitrogen Legend:	Not Available Extracted from Ecotox databas	Not Available 1. IUCLID Toxicity Data 2. Europe ECH.	·	Not Available ation - Aquatic Toxicity 4.	υ

May cause long-term adverse effects in the aquatic environment.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
ammonium phosphate, monobasic	нівн	HIGH
ammonium sulfate	HIGH	HIGH

Bioaccumulative potential

Ingredient	Bioaccumulation
ammonium phosphate, monobasic	LOW (LogKOW = -0.7699)
ammonium sulfate	LOW (LogKOW = -2.2002)

Mobility in soil

Ingredient	Mobility
ammonium phosphate, monobasic	HIGH (KOC = 1)
ammonium sulfate	LOW (KOC = 6.124)

SECTION 13 Disposal considerations

Waste treatment methods

Product / Packaging disposal

- ► Evaporate residue at an approved site.
- Return empty containers to supplier. If containers are marked non-returnable establish means of disposal with manufacturer prior to purchase.
- ► Ensure damaged or non-returnable cylinders are gas-free before disposal.

SECTION 14 Transport information

Labels Required

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Marine Pollutant

HAZCHEM Not Applicable

Land transport (ADG)

UN number	1044	
UN proper shipping name	FIRE EXTINGUISHERS with compressed or liquefied gas	
Transport hazard class(es)	Class 2.2 Subrisk Not Applicable	
Packing group	Not Applicable	
Environmental hazard	Not Applicable	
Special precautions for user	Special provisions 225 Limited quantity 120 ml	

Air transport (ICAO-IATA / DGR)

Air transport (ICAO-IAIA / DGR)				
UN number	1044			
UN proper shipping name	Fire extinguishers with compressed or liquefied gas			
Transport hazard class(es)	ICAO/IATA Class ICAO / IATA Subrisk ERG Code	2.2 Not Applicable 2L		
Packing group	Not Applicable			
Environmental hazard	Not Applicable			
Special precautions for user	Special provisions		A19	
	Cargo Only Packing Instructions		213	
	Cargo Only Maximum Qty / Pack		150 kg	
	Passenger and Cargo Packing Instructions		213	
	Passenger and Cargo Maximum Qty / Pack		75 kg	
	Passenger and Cargo Limited Quantity Packing Instructions		Forbidden	
	Passenger and Cargo Limited Maximum Qty / Pack		Forbidden	

Sea transport (IMDG-Code / GGVSee)

UN number	1044		
UN proper shipping name	FIRE EXTINGUISHERS with compressed or liquefied gas		
Transport hazard class(es)	IMDG Class 2.2 IMDG Subrisk Not Applicable		
Packing group	Not Applicable		
Environmental hazard	Not Applicable		
Special precautions for user	Special provisions	F-C, S-V 225 120 mL	

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
ammonium phosphate, monobasic	Not Available
ammonium sulfate	Not Available
staurolite	Not Available
nitrogen	Not Available

Transport in bulk in accordance with the ICG Code

Product name Ship Type Chemwatch: 5570-64 Page 10 of 11 Issue Date: 13/01/2023 Version No: 2.1 Print Date: 13/01/2023

ABE Powder Extinguisher with compressed gas

Product name	Ship Type
ammonium phosphate, monobasic	Not Available
ammonium sulfate	Not Available
staurolite	Not Available
nitrogen	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

ammonium phosphate, monobasic is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

ammonium sulfate is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

FEI Equine Prohibited Substances List - Banned Substances

FEI Equine Prohibited Substances List (EPSL)

staurolite is found on the following regulatory lists

International WHO List of Proposed Occupational Exposure Limit (OEL) Values for Manufactured Nanomaterials (MNMS)

nitrogen is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

National Inventory Status

National Inventory	Status	
Australia - AIIC / Australia Non-Industrial Use	No (staurolite)	
Canada - DSL	No (staurolite)	
Canada - NDSL	No (ammonium phosphate, monobasic; ammonium sulfate; staurolite; nitrogen)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	No (staurolite)	
Japan - ENCS	No (staurolite; nitrogen)	
Korea - KECI	No (staurolite)	
New Zealand - NZIoC	No (staurolite)	
Philippines - PICCS	Yes	
USA - TSCA	No (staurolite)	
Taiwan - TCSI	Yes	
Mexico - INSQ	No (staurolite)	
Vietnam - NCI	No (staurolite)	
Russia - FBEPH	No (staurolite)	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

SECTION 16 Other information

Revision Date	13/01/2023
Initial Date	13/01/2023

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

ES: Exposure Standard

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors Chemwatch: 5570-64 Page 11 of 11 Issue Date: 13/01/2023 Version No: 2.1 Print Date: 13/01/2023

ABE Powder Extinguisher with compressed gas

BEI: Biological Exposure Index

AIIC: Australian Inventory of Industrial Chemicals

DSL: Domestic Substances List

NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances

ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers

ENCS: Existing and New Chemical Substances Inventory

KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act

TCSI: Taiwan Chemical Substance Inventory

INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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